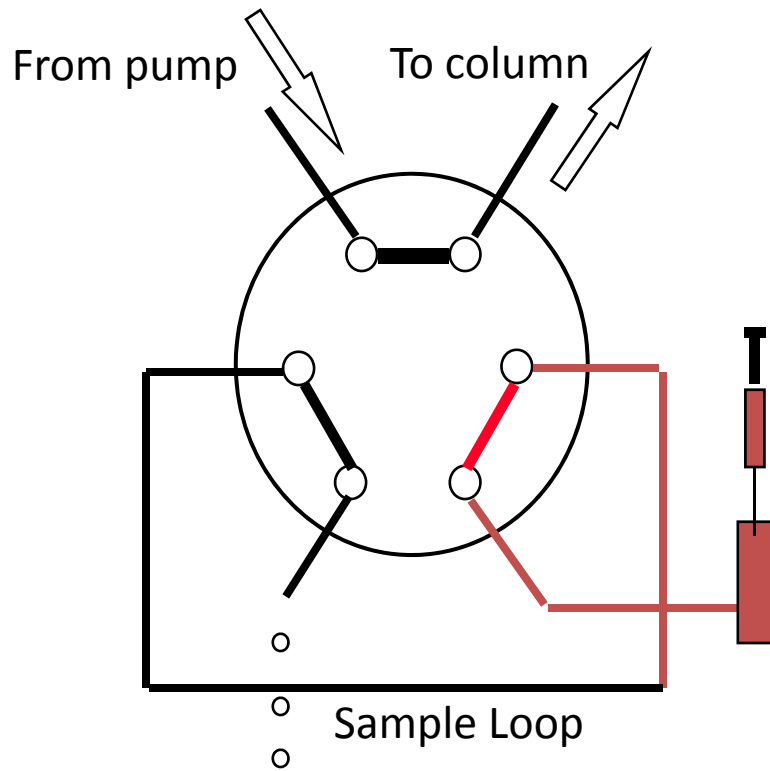


# **Semester VI (CHB-601)**

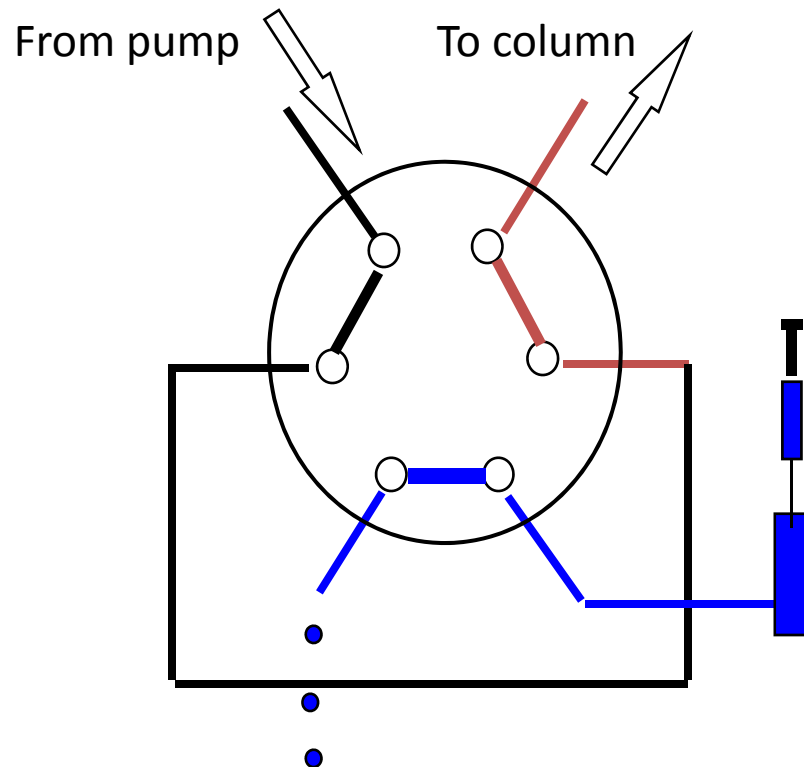
## **Separation of molecules by chromatographic techniques**

Dr Meenakshi Singh  
Professor of Chemistry  
Mahila Mahavidyalaya  
Banaras Hindu University  
Varanasi, INDIA

# Autosampler (Pressure Injection Method)

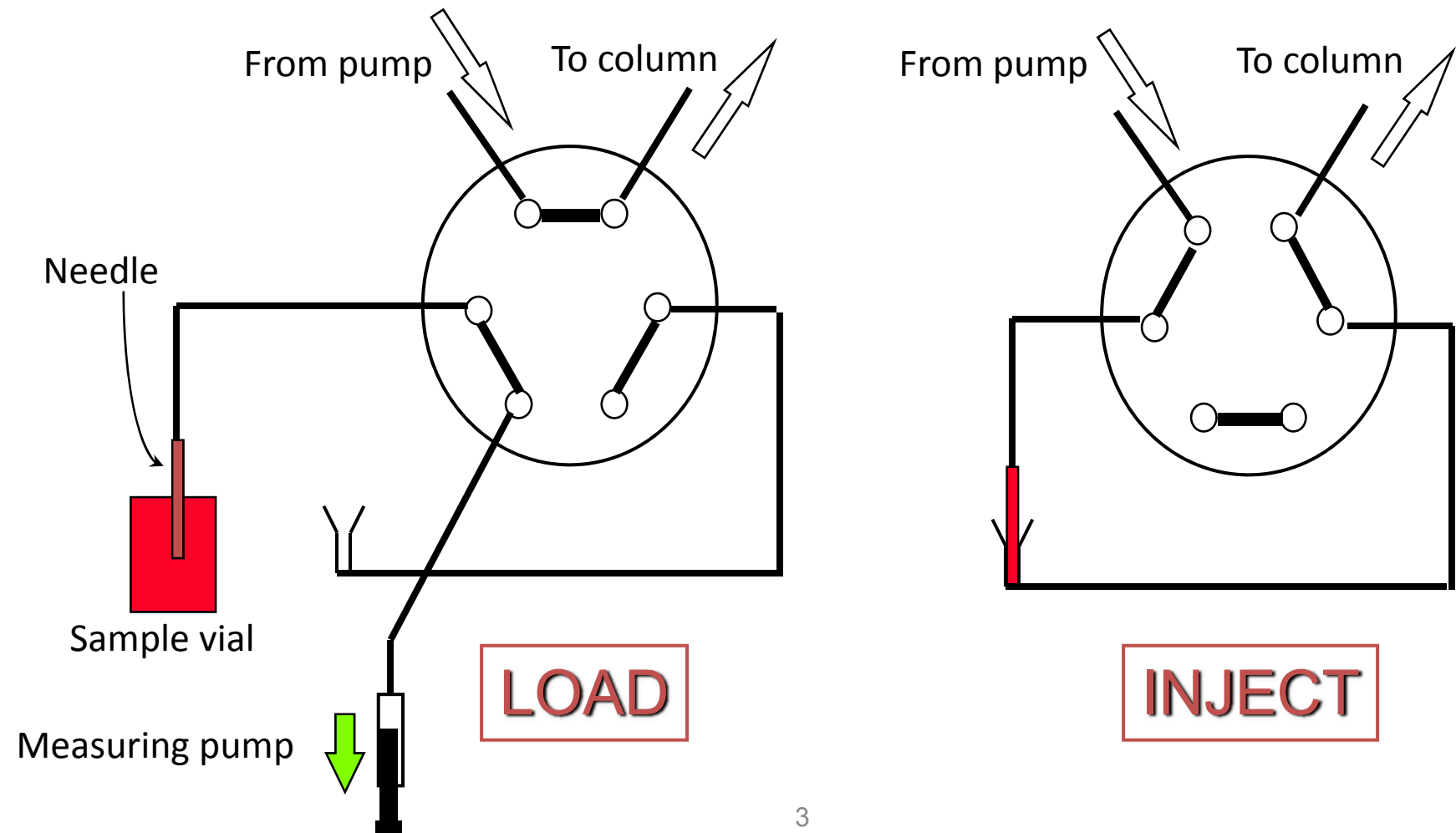


**LOAD**



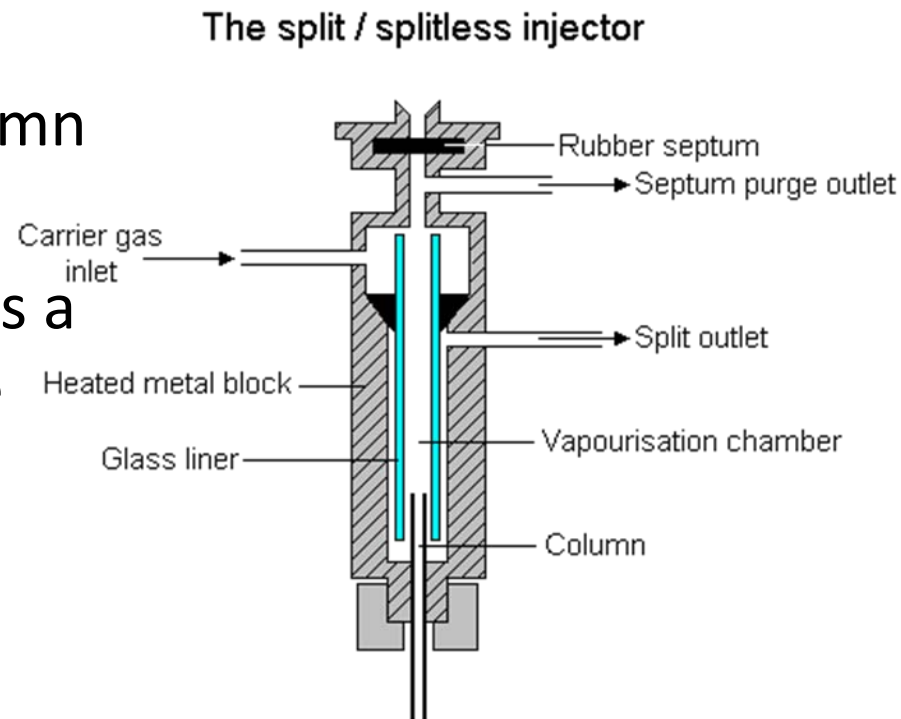
**INJECT**

# Autosampler (Total-Volume Injection Method)

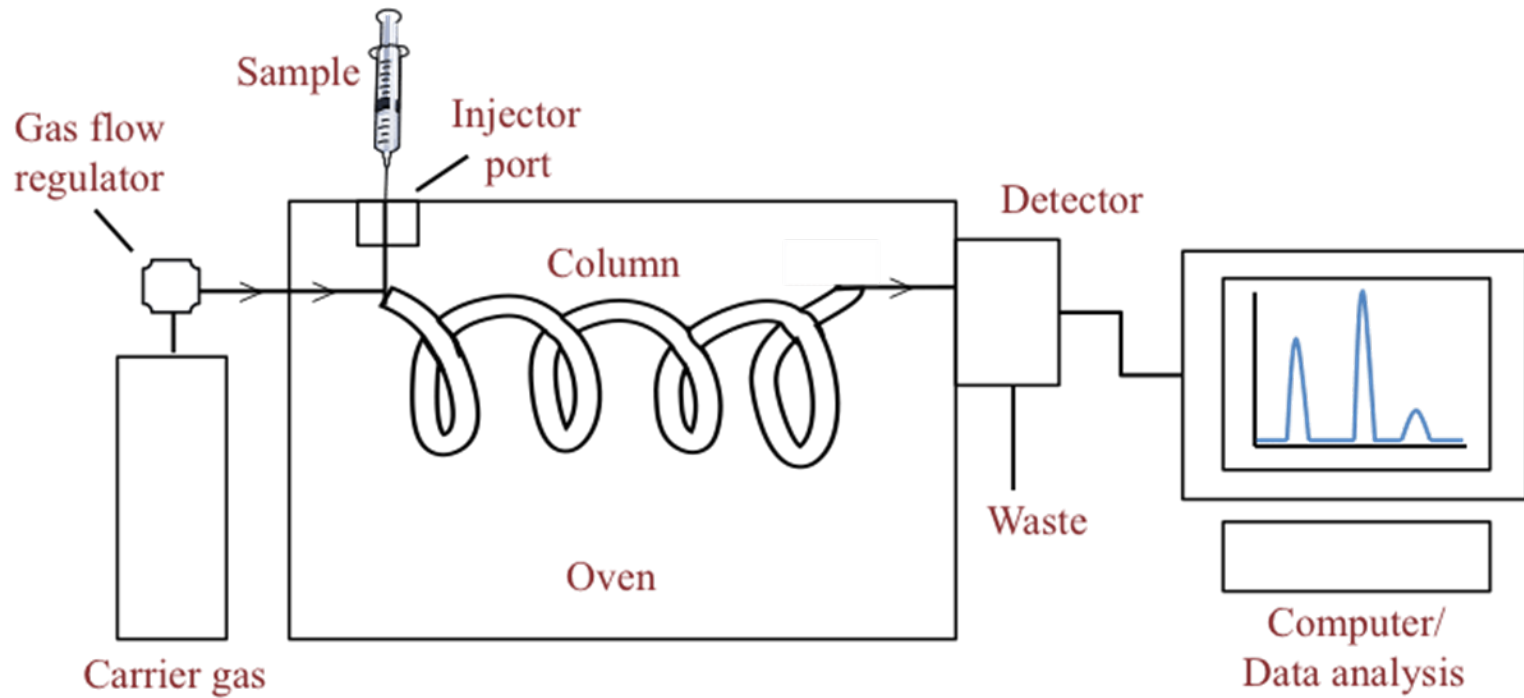


# Gas Chromatography

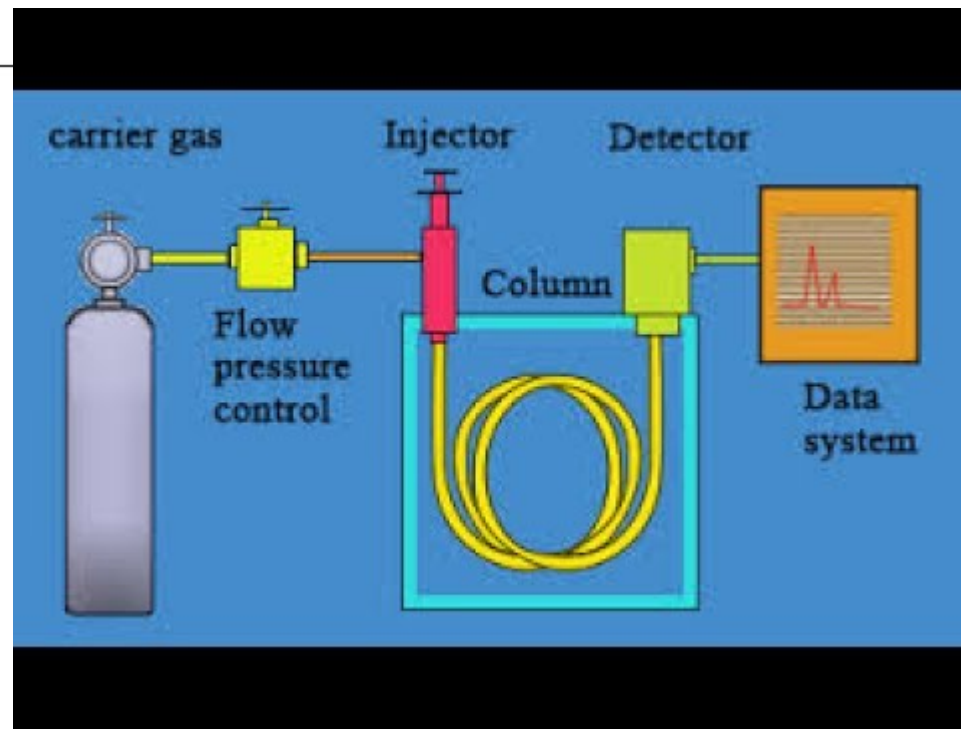
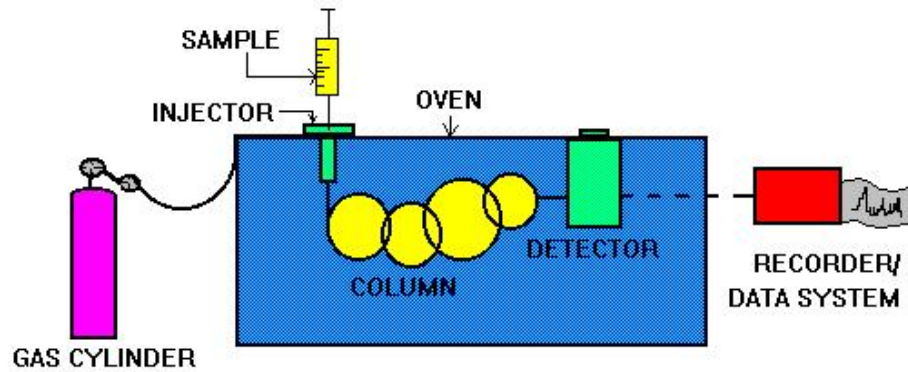
- A vaporized sample is injected onto the chromatographic column
- Sample moves through the column through the flow of inert gas
- The components are recorded as a sequence of peaks as they leave the column
- Mobile phase-inert gas used as carrier
- Stationary phase- liquid coated on a solid or solid within a column



# Apparatus outline



# GAS CHROMATOGRAPHY





# Practical requirements

- Carrier gas
- Flow regulators and flow meters
- Injection devices
- Columns
- Temperature control devices
- Detectors
- Recorders and integrators



# Carrier gas

- Hydrogen
  - Better thermal conductivity
  - Disadvantages
    - Reacts with unsaturated compounds
    - Inflammable
- Helium
  - excellent thermal conductivity
  - Expensive
- Nitrogen
  - Reduced sensitivity
  - inexpensive

# Requirements of a carrier gas

- Inertness
- Suitable for detector
- High purity
- Easily available
- Cheap
- Should give best column performance

# Ion exchange chromatography

## Principle

- ▶ Ion exchange chromatography retains analyte molecules based on ionic interactions.
  - ▶ The stationary phase surface displays ionic functional groups (R-X) that interact with analyte ions of opposite charge.
- 

